Table 6. Torsion angles $\left(^{\circ}\right)$ describing the conformation of the molecule

| Theophylline | Pyranose ring |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(4)$ | 3.0 (3) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 11.1(2) | $\mathrm{H}\left(3^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | -43.2 (1) |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | -4.0 (3) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | $6 \cdot 6$ (2) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)$ | -167.5 (1) |
| $\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)$ | 2.6 (3) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 11.4 (2) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | 72.7 (1) |
| $\mathrm{C}(4)-\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{N}(1)$ | 0.0 (3) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)$ | -47.0 (2) | $\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)$ | 77.1 (2) |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(3)-\mathrm{C}(2)$ | -1.0 (3) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)$ | 68.4 (2) | $\mathrm{H}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | -46.2 (1) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(3)$ | 0.0 (3) | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | -48.8 (2) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)$ | 82.5 (2) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}(8)$ | 0.0 (2) | Others |  | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)$ | -147.3 (2) |
| $\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{N}(9)$ | $0 \cdot 0$ (2) |  |  | $\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{C}\left(11^{\prime}\right)$ | -170.8(2) |
| $\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{N}(9)-\mathrm{C}(4)$ | $0 \cdot 0$ (2) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)$ | -169.2 (1) | $\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | $7 \cdot 8$ (3) |
| $\mathrm{C}(8)-\mathrm{N}(9)-\mathrm{C}(4)-\mathrm{C}(5)$ | $0 \cdot 0$ (2) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | -111.9 (2) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(7^{\prime}\right)$ | 67.1 (2) |
| $\mathrm{N}(9)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(7)$ | $0 \cdot 0$ (2) | $\mathrm{H}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 127.1 (2) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)$ | -170.9 (2) |
|  |  | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{H}\left(3^{\prime}\right)$ | -180.0 (2) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | -8.6(3) |
|  |  | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{N}(7)$ | -113.1 (2) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{C}\left(9^{\prime}\right)$ | 170.4 (2) |
|  |  | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | 130.4 (1) | $\mathrm{C}(8)-\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | -81.0 (2) |

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# The Crystal and Molecular Structure of 7-(Methyl 2-acetamido-6-O-acetyl-2,3,4-trideoxy-$\beta$-D-erythro-hex-2-enopyranosid-4-yl)theophylline, $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{7}$ 

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#### Abstract

The title compound crystallizes in the monoclinic space group $C 2$ with $a=24.093$ (1), $b=5.003$ (3), $c=$ 19.002 (2) $\AA, \beta=118.85(4)^{\circ}, Z=4$. The structure was refined to an $R$ of 0.051 . The orientation of the base relative to the sugar ring, defined in terms of rotation about the $\mathrm{C}\left(4^{\prime}\right)-\mathrm{N}(7)$ glycosyl bond, is syn 0567-7408/79/051166-05\$01.00


$\left(105 \cdot 1^{\circ}\right)$. The theophylline base is planar within the limits of experimental error. The sugar moiety exhibits a half-chair ${ }^{5} H_{0}$ conformation. The molecules are connected by $\mathrm{N}\left(1^{\prime}\right)-\mathrm{H} \cdots \mathrm{O}(2), 3.09 \AA$, hydrogen bonds acting between the sugar and base moieties forming an infinite chain along a. Base stacking occurs along $\mathbf{b}$.
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## Experimental

The space group was determined from Weissenberg photographs recorded with $\mathrm{Cu} K a$ radiation. The diffraction symmetry and space-group extinctions indicated $C 2$ or $C m$; the structure was solved and refined successfully in C2. Table 1 lists the crystallographic and physical data.

The intensities were collected on a Philips PW 1100 computer-controlled four-circle diffractometer in the $\omega$ scan mode [scan width $=1 \cdot 20^{\circ}(\theta)$, scan speed $=$ $0.03^{\circ}(\theta) \mathrm{s}^{-1}$ ] with graphite-monochromated $\mathrm{Cu} K \alpha$ radiation. 1270 independent reflexions $[I>2 \sigma(I)$ ] in the range $2<\theta<70^{\circ}$ were used in the calculations. Three standard reflexions were measured every 2 h . The data were corrected for background, Lorentz and polarization effects but not for absorption.

## Structure determination and refinement

An overall temperature factor ( $B=2.04 \AA^{2}$ ) and a scale factor were determined (Wilson, 1942) and used to compute normalized structure amplitudes by the routine NORMAL included in MULTAN. The geometry of the planar molecular fragment ( 11 atoms of the theophylline base) was used to modify the Wilson plot by applying the spherically averaged molecular scattering factor. 300 reflexions with $|E|>1 \cdot 15$ were used. MULTAN (Main, Lessinger, Woolfson, Germain \& Declercq, 1977) failed; no set of phases yielded an $E$ map in which all or even a recognizable fragment appeared. Particular tactics described by Lessinger (1976) were applied but without success. The MULTAN failure could be attributed to the relatively small number of recorded intensities and difficulties typical of the $C 2$ space group. To find a solution a magic-integer representation for a symbolic phase assignment with the $P, S$ method (Woolfson, 1976; Main, 1977) was employed to derive phases. The values for 42 phases were obtained and put as known phases into MULTAN (Main et al., 1977). $4000 \sum_{2}$ phase relationships were used. The relative weights for ABS FOM, PSI ZERO and RESID were used; ABS FOM

## Table 1. Crystallographic and physical data

Numbers in parentheses here and throughout this paper are the e.s.d.'s in the least significant digits.

| 7-(Methyl |  |  |  |
| :--- | :--- | :--- | :--- |
| 2-acetamido-6- $O$-acetyl-2,3,4-trideoxy- $\beta$-D-erythro-hex-2- |  |  |  |
| enopyranosid-4-yl) | theophylline, $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}$, |  |  |
|  |  |  |  |
| FW | 437.42 | $U$ | $2006 \cdot 17 \AA^{3}$ |
| Space group | $C 2$ | $Z$ | 4 |
| $a$ | $24.093(1) \AA$ | $D_{c}$ | $1.407 \mathrm{Mg} \mathrm{m}^{-3}$ |
| $b$ | $5.003(3)$ | $\mu(\mathrm{Cu} \mathrm{Ka})$ | $1.137 \mathrm{~mm}^{-1}$ |
| $c$ | $19.002(2)$ | Crystal shape | Needle |
| $\beta$ | $118.85(4)^{\circ}$ | Crystal size | $0.028 \times 0.042 \times 0.566 \mathrm{~mm}$ |

was given less and PSI ZERO more weight. The set with ABS FOM $=1 \cdot 304$, PSI ZERO $=33 \cdot 2$, RESID $=31.44$ was selected and revealed 29 nonhydrogen atoms out of 30 in the molecule. The same fragment occurred in the $E$ map using 15 known phases with $|E|>2.0$ in the MULTAN input. The remaining C atom belonging to the terminal methyl group was located from the resulting Fourier synthesis.

Refinement was by a full-matrix least-squares method minimizing $\sum w\left|F_{o}\right|-\left|F_{c}\right|^{2}$ with $w=1 / \sigma_{F_{s}}^{2}$. Anisotropic refinement and a subsequent difference synthesis were performed to locate all H atoms. In the final cycles one scale factor, and the positional and anisotropic thermal parameters ( 269 in all) were varied. The H atoms were included only in the structure factor calculations. For the H atoms, the isotropic thermal parameters were those of the bonded atom. The final $R=0.051$ and $R_{w}=0.064$. The analysis of the title compound is of limited accuracy because of the poor quality of the crystal.
Scattering factors given by Cromer \& Mann (1968) and (for H) Stewart, Davidson \& Simpson (1965) were used.

The calculations were performed on a Univac 1110 computer at the University Computing Centre in Zagreb with the XRAY system (Stewart, Kruger, Ammon, Dickinson \& Hall, 1972).

Table 2. Final atomic parameters $\left(\times 10^{4}\right)$ for the non-hydrogen atoms

|  |  |  |  |
| :--- | ---: | ---: | ---: |
|  |  | $z$ |  |
| $\mathrm{C}(1)$ | $2556(3)$ | $-1777(23)$ | $3120(5)$ |
| $\mathrm{C}(2)$ | $1803(3)$ | $-4451(16)$ | $1969(4)$ |
| $\mathrm{C}(3)$ | $1054(3)$ | $-6985(19)$ | $780(4)$ |
| $\mathrm{C}(4)$ | $702(3)$ | $-3822(16)$ | $1472(4)$ |
| $\mathrm{C}(5)$ | $805(3)$ | $-1912(16)$ | $2023(4)$ |
| $\mathrm{C}(6)$ | $1420(3)$ | $-1075(0)$ | $2599(4)$ |
| $\mathrm{C}(8)$ | $-212(3)$ | $-2523(16)$ | $1231(4)$ |
| $\mathrm{C}\left(1^{\prime}\right)$ | $-924(3)$ | $-1160(15)$ | $2657(4)$ |
| $\mathrm{C}\left(2^{\prime}\right)$ | $-1063(3)$ | $953(16)$ | $2019(4)$ |
| $\mathrm{C}\left(3^{\prime}\right)$ | $-627(3)$ | $1801(16)$ | $1828(4)$ |
| $\mathrm{C}\left(4^{\prime}\right)$ | $48(3)$ | $915(16)$ | $2325(4)$ |
| $\mathrm{C}\left(5^{\prime}\right)$ | $186(3)$ | $-215(15)$ | $3127(4)$ |
| $\mathrm{C}\left(6^{\prime}\right)$ | $245(3)$ | $1971(16)$ | $3729(4)$ |
| $\mathrm{C}\left(7^{\prime}\right)$ | $-1054(4)$ | $-1963(23)$ | $3786(5)$ |
| $\mathrm{C}\left(8^{\prime}\right)$ | $-1967(3)$ | $3639(16)$ | $1039(4)$ |
| $\mathrm{C}\left(9^{\prime}\right)$ | $-2657(3)$ | $4170(20)$ | $747(4)$ |
| $\mathrm{C}\left(10^{\prime}\right)$ | $1351(3)$ | $3016(16)$ | $4459(4)$ |
| $\mathrm{C}\left(11^{\prime}\right)$ | $1845(3)$ | $4965(18)$ | $4531(5)$ |
| $\mathrm{N}(1)$ | $1895(2)$ | $-2500(15)$ | $2530(3)$ |
| $\mathrm{N}(3)$ | $1191(2)$ | $-5083(14)$ | $1426(3)$ |
| $\mathrm{N}(7)$ | $205(2)$ | $-1044(12)$ | $1869(3)$ |
| $\mathrm{N}(9)$ | $74(2)$ | $-4277(15)$ | $975(3)$ |
| $\mathrm{N}\left(1^{\prime}\right)$ | $-1708(2)$ | $1688(13)$ | $1625(3)$ |
| $\mathrm{O}(2)$ | $2246(2)$ | $-5596(12)$ | $1949(3)$ |
| $\mathrm{O}(6)$ | $1557(2)$ | $720(0)$ | $3099(3)$ |
| $\mathrm{O}\left(0^{\prime}\right)$ | $-300(2)$ | $-2104(10)$ | $3014(2)$ |
| $\mathrm{O}\left(1^{\prime}\right)$ | $-1071(2)$ | $26(12)$ | $3228(3)$ |
| $\mathrm{O}\left(2^{\prime}\right)$ | $-1674(2)$ | $4803(14)$ | $771(3)$ |
| $\mathrm{O}\left(3^{\prime}\right)$ | $776(2)$ | $3679(11)$ | $3866(3)$ |
| $\mathrm{O}\left(4^{\prime}\right)$ | $1444(2)$ | $1067(14)$ | $4878(3)$ |
|  |  |  |  |

Atomic positional parameters are listed in Tables 2 and 3.*

## Description and discussion of the structure

The present structure determination is part of an investigation of the new type of $2^{\prime}, 3^{\prime}$-unsaturated nucleosides with the base linked at $\mathbf{C}\left(4^{\prime}\right)$ (Kojić-Prodić, Danilov \& Pravdić, 1976; Kojić-Prodić, 1979; Pravdić, 1979).

> * Lists of structure factors and thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34188 (15 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 3. Positional parameters $\left(\times 10^{3}\right)$ for the hydrogen atoms

|  | $x$ | $y$ | $z$ |
| :---: | :---: | :---: | :---: |
| H(1)1 | 258 | 000 | 350 |
| H(1)2 | 267 | -350 | 325 |
| H(1)3 | 267 | -70 | 275 |
| H(3)1 | 117 | -900 | 100 |
| H(3)2 | 133 | -733 | 67 |
| H(3)3 | 58 | -767 | 50 |
| H(8) | -65 | -233 | 92 |
| H(1') | -125 | -280 | 242 |
| H(3') | -81 | 300 | 125 |
| H(4) | 33 | 260 | 242 |
| H(5) | 58 | -167 | 342 |
| H(6) 1 | -15 | 333 | 350 |
| H(6) 2 | 42 | 100 | 433 |
| H(7) 1 | -137 | -367 | 367 |
| $\mathrm{H}\left(7^{\prime}\right) 2$ | -56 | -267 | 417 |
| H( $7^{\prime}$ ) 3 | -100 | -40 | 425 |
| H(9')1 | -281 | 247 | 100 |
| H(9')2 | -281 | 500 | 25 |
| H(9')3 | -268 | 533 | 120 |
| H(11) 1 | 195 | 367 | 413 |
| H(11')2 | 217 | 533 | 510 |
| H(11')3 | 167 | 667 | 425 |
| H( $\mathrm{l}^{\prime}$ ) | -212 | 100 | 170 |



Fig. 1. The structural formula and intramolecular distances ( $\AA$ ).

The structural formula with the atom numbering and interatomic distances is shown in Fig. 1. Bond angles are listed in Table 4. The molecular packing and hydrogen bonds are illustrated in Fig. 2. The conformation of the carbohydrate moiety is shown in Fig. 3 and Tables 5 and 6. Torsion angles defining the conformation of the molecule are presented in Table 6.

Table 4. Bond angles $\left({ }^{\circ}\right)$

| (1)-C(2) | 118.0 (6) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 112.9 (6) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(1)-\mathrm{N}(1)-\mathrm{C}(6)$ | 115.5 (6) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{N}(7)$ | 109.6 (5) |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)$ | 126.5 (4) | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{N}(7)$ | 110.7 (6) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{O}(2)$ | 122.3 (5) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | 109.3 (4) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{N}(3)$ | 117.3 (6) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)$ | 110.9 (6) |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{N}(3)$ | 120.4 (7) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)$ | 112.4 (6) |
| $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{C}(3)$ | $120 \cdot 6$ (6) | $\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 108.1 (6) |
| $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{C}(4)$ | 119.4 (6) | $\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | 123.7 (6) |
| $\mathrm{C}(3)-\mathrm{N}(3)-\mathrm{C}(4)$ | 120.0 (5) | $\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{C}\left(9^{\prime}\right)$ | 114.3 (7) |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 122.2 (5) | $\mathrm{O}\left(2^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{C}\left(9^{\prime}\right)$ | 122.0 (7) |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{N}(9)$ | 125.3 (6) | $\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | 122.7 (7) |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(9)$ | 112.5 (6) | $\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{C}\left(11^{\prime}\right)$ | 111.7 (6) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(7)$ | 105.9 (5) | $\mathrm{O}\left(4^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{C}\left(11^{\prime}\right)$ | 125.5 (5) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 122.7 (6) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(7^{\prime}\right)$ | $110 \cdot 2$ (7) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{N}(7)$ | 131.4 (7) | $\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)$ | 118.0 (6) |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | 112.0 (5) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)$ | 124.4 (6) |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{O}(6)$ | 120.9 (5) | $\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{H}(8)$ | 129 |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{O}(6)$ | 127.1 (6) | $\mathrm{N}(9)-\mathrm{C}(8)-\mathrm{H}(8)$ | 117 |
| $\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}(8)$ | 105.2 (6) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{H}\left(\mathrm{N} 1^{\prime}\right)$ | 132 |
| $\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)$ | 128.0 (5) | $\mathrm{C}\left(8^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{H}\left(\mathrm{N} 1^{\prime}\right)$ | 104 |
| $\mathrm{C}(8)-\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)$ | 126.6 (5) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{H}\left(1^{\prime}\right)$ | 110 |
| $\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{N}(9)$ | 112.8 (5) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{H}\left(1^{\prime}\right)$ | 104 |
| $\mathrm{C}(4)-\mathrm{N}(9)-\mathrm{C}(8)$ | 103.3 (6) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{H}\left(1^{\prime}\right)$ | 111 |
| $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 117.2 (5) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{H}\left(3^{\prime}\right)$ | 116 |
| $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)$ | 112.0 (4) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{H}\left(3^{\prime}\right)$ | 124 |
| $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | 113.1 (6) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | 107 |
| $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $106 \cdot 2$ (6) | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | 109 |
| $\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)$ | 111.6 (6) | $\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | 108 |
| $\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 125.7 (6) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | 95 |
| $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 122.5 (6) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | 118 |
| $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | 119.5 (6) | $\mathrm{C}\left(6^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | 110 |



Fig. 2. Molecular packing viewed along b. Intermolecular hydrogen bonds between the amide H atom (in the sugar moiety) and the keto O atom (in the base moiety) are represented by dotted lines The $\mathrm{N}\left(1^{\prime}\right) \cdots \mathrm{O}(2)$ distance is $3.09 \AA$. A right-handed coordinate system has been chosen so that the positive direction of $\mathbf{b}$ is away from the reader.

In the preparation of the title compound, the $D$ compound was used as the starting material. In the present structure determination the $D$ enantiomer was selected and the configuration and conformation were defined in accordance with its torsion angles (Table 6).

Base stacking is shown in Fig. 4.

## Base

The theophylline base (1,3-dimethylxanthine) realizes the glycosyl bond through the $\mathrm{N}(7) \mathrm{H}$ form. The e.s.d.'s of bond distances and angles are much


Fig. 3. Schematic drawing of the ${ }^{5} H_{0}$ half-chair conformation illustrating the displacements of $\mathrm{C}\left(5^{\prime}\right)$ (above) and $\mathrm{O}\left(0^{\prime}\right)$ (below) the plane defined by $\mathrm{C}\left(1^{\prime}\right), \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(3^{\prime}\right)$ and $\mathrm{C}\left(4^{\prime}\right)$, and the orientation of the substituents.

Table 5. Displacements from least-squares planes through the base and sugar ( $\AA$ )

Atoms included in the calculation of the planes are denoted by an asterisk.

larger than those for the $\alpha$-D-threo isomer (KojićProdić, 1979).

The base conformation is defined by the mean torsion angles of 1 and $0.8^{\circ}$ for the six- and fivemembered rings, respectively (Table 6). The base is almost planar; the displacements of the atoms from the least-squares planes (Table 5) support this fact.

## Sugar moiety

Bond lengths and angles (Fig. 1 and Table 4) are comparable with the values found in the $\alpha$-D-threoisomer (Kojić-Prodic, 1979) and also in some peracetylated 2,3-dideoxyaldopyranoses (Rogić, 1975; Kojić-Prodić, Rogić \& Ružić-Toroš, 1976). Lower accuracy of the data prevents us from drawing any conclusion on the $\mathrm{C}-\mathrm{O}$ distances associated with the anomeric $C$ atom. The endocyclic $C\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)$ bond angle of $117 \cdot 2^{\circ}$ is rather large although a value of $116 \cdot 1^{\circ}$ was found in sucrose ( $\alpha$-D-glucose residue) (Hybl, Rundle \& Williams, 1965).

The carbohydrate moiety exhibits the ${ }^{5} H_{0}$, half-chair conformation (Fig. 3), possible for $\beta$-anomers (Stoddart, 1971). The best least-squares plane is defined by $C\left(1^{\prime}\right), C\left(2^{\prime}\right), C\left(3^{\prime}\right)$ and $C\left(4^{\prime}\right) ; C\left(5^{\prime}\right)$ and $O\left(0^{\prime}\right)$ are displaced from this plane by 0.532 and $-0.141 \AA$, respectively (Table 5). The ring substituents are attached at $C\left(1^{\prime}\right)$ and $C\left(4^{\prime}\right)$ in quasi-axial and $C\left(5^{\prime}\right)$ in


Fig. 4. Base stacking along $\mathbf{b}$ with the shortest interbase distance of $5.003 \AA$.

Table 6. Torsion angles $\left({ }^{\circ}\right)$ describing the conformation of the molecule

| Theophylline | Pyranose ring |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(4)$ | -1 (1) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 0 (1) | $\mathrm{H}\left(3^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | 50 |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | 2 (1) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | -8(1) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)$ | -75.8 (8) |
| $\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)$ | 0 (1) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | -17(1) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | 154 |
| $\mathrm{C}(4)-\mathrm{N}(3)-\mathrm{C}(2)-\mathrm{N}(1)$ | -2 (1) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)$ | 47.7 (8) | $\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)$ | -75.6 (7) |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(3)-\mathrm{C}(2)$ | 2 (1) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)$ | -59.3 (8) | $\mathrm{H}\left(4^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{H}\left(5^{\prime}\right)$ | -87 |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{N}(3)$ | -1 (1) | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $35 \cdot 3$ (8) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)$ | 175.3 (5) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}(8)$ | -0.1 (8) | Others |  | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)$ | -89.4 (7) |
| $\mathrm{C}(5)-\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{N}(9)$ | $0 \cdot 8$ (8) |  |  | $\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{C}\left(11^{\prime}\right)$ | -179.5 (7) |
| $\mathrm{N}(7)-\mathrm{C}(8)-\mathrm{N}(9)-\mathrm{C}(4)$ | -1.2(8) | $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)$ | 174.9 (6) | $\mathrm{C}\left(6^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(10^{\prime}\right)-\mathrm{O}\left(4^{\prime}\right)$ | 0 (1) |
| $\mathrm{C}(8)-\mathrm{N}(9)-\mathrm{C}(4)-\mathrm{C}(5)$ | 1.1 (9) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 123.0 (8) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(7^{\prime}\right)$ | -65.6 (7) |
| $\mathrm{N}(9)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{N}(7)$ | -0.7 (9) | $\mathrm{H}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | -125 | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)$ | 177.4 (7) |
|  |  | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{H}\left(3^{\prime}\right)$ | 166 | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{O}\left(2^{\prime}\right)$ | 4 (1) |
|  |  | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{N}(7)$ | 107.2 (8) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{N}\left(1^{\prime}\right)-\mathrm{C}\left(8^{\prime}\right)-\mathrm{C}\left(9^{\prime}\right)$ | -176.5 (7) |
|  |  | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{H}\left(4^{\prime}\right)$ | -136 | $\mathrm{C}(8)-\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | $105 \cdot 1$ (7) |

axial positions (Fig. 3) (Stoddart, 1971), as expected for the ${ }^{5} H_{0}$ half-chair conformation. The highly distorted conformation appears to be due to the departure of $\mathrm{C}\left(5^{\prime}\right)$ from the plane to avoid steric hindrance between bulky substituents at the contiguous $C\left(4^{\prime}\right)$ and $C\left(5^{\prime}\right)$.

The puckering of the sugar ring required by the ${ }^{5} H_{0}$ conformation can be described by the values of the $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)\left(-17^{\circ}\right)$ and $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-$ $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(0^{\prime}\right)\left(47 \cdot 7^{\circ}\right)$ torsion angles (Table 6).

Sundaralingam (1968) has defined the conformation about $\mathrm{C}(5)-\mathrm{C}(6)$ in pyranosides by the angle $\varphi_{\mathrm{oo}}=$ $\mathrm{O}(5)-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{O}(6)$. In the present structure this angle is described by the squence $\mathrm{O}\left(0^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{C}\left(6^{\prime}\right)-$ $O\left(3^{\prime}\right)$ and its value of $175.3^{\circ}$ is not in the range common to pyranoside derivatives ( $\pm 60 \pm 30^{\circ}$ ).

## Nucleoside conformation and molecular packing

The orientation of the base relative to the sugar ring described in terms of rotation about the $\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)$ glycosyl bond for the sequence $\mathrm{C}(8)-\mathrm{N}(7)-\mathrm{C}\left(4^{\prime}\right)-$ $\mathrm{C}\left(5^{\prime}\right)$ is syn ( $105 \cdot 1^{\circ}$ ) (Sundaralingam, 1975). The conformation about the glycosyl bond with the torsion angle ( $\chi$ ) larger than $90^{\circ}$ can also be referred to as high-anti ( $90-130^{\circ}$ ) (Sundaralingam, 1975). Thus, a comparison of the molecules in the usual anti range can be made.

The molecules are joined by $\mathrm{N}\left(1^{\prime}\right)-\mathrm{H} \cdots \mathrm{O}(2)$ (3.09 $\AA$ ) hydrogen bonds acting between the sugar and base moieties (Fig. 2). Each molecule is connected by two hydrogen bonds with the neighbouring molecule forming infinite chains along a.

Base stacking with the shortest interbase distance of $5.003 \AA$ occurs along b.

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# The Crystal Structures of Two Fluorene Derivatives 

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#### Abstract

The crystal structures of 2,7-dibromo-9-[4-(dimethylamino) benzylidene]fluorene (I) ( $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{Br}_{2} \mathrm{~N}$ ) and 9,9'-(1,4-phenylenedimethylidyne)difluorene (II) $\left(\mathrm{C}_{34} \mathrm{H}_{22}\right)$ were determined by X-ray structure analysis. The


 0567-7408/79/051170-05\$01.00crystals are monoclinic, space group $P 2_{1} / c$, with $a=$ 8.200 (6), $b=14.536$ (7), $c=15.729$ (5) $\AA, \beta=$ 103.64 (6) $^{\circ}$ and $Z=4$ for (I), and orthorhombic, space group $P b c a$, with $a=17.640$ (6), $b=23 \cdot 630$ (6), $c=10.755$ (5) $\AA$ and $Z=8$ for (II). The structures were solved by Patterson methods for (I) and direct © 1979 International Union of Crystallography

